

Useful Results from Statistical Thermodynamics

$$U = kT \left(\frac{\partial \ln Q}{\partial \ln T} \right)_{V,N} = kT^2 \left(\frac{\partial \ln Q}{\partial T} \right)_{V,N}$$

$$S = k \left[\ln Q + \left(\frac{\partial \ln Q}{\partial \ln T} \right)_{V,N} \right]$$

$$p = kT \left(\frac{\partial \ln Q}{\partial V} \right)_{T,N}$$

Ideal Gas

Consider translational, rotational, vibrational, and electronic parts. Treat electronic parts on a case-by-case basis. Often, molecules are electronically in their ground states. Otherwise, we need to know the energy levels corresponding to other states.

Partition Function:

$$q_t = \left[\frac{2\pi kT \sum m_i}{h^2} \right]^{3/2} V$$

$$\text{polyatomic: } q_r = \frac{\pi^{1/2}}{\sigma} \left(\frac{8\pi^2 kT}{h^2} \right)^{3/2} (I_A I_B I_C)^{1/2}$$

$$\text{diatomic: } q_r = \frac{8\pi^2 I kT}{\sigma h^2}$$

$$q_v = \prod_{i=1}^{n'} \frac{\exp(-\Theta_{v,i} / 2T)}{1 - \exp(-\Theta_{v,i} / T)} \quad n' = 3n-5 \text{ (linear or diatomic) or } 3n-6 \text{ (non-linear)}$$

$$\Theta_{v,i} = h \nu_i / k$$

Internal Energy:

$$E_t = \frac{3}{2} NkT$$

$$E_r = NkT, \text{ linear or diatomic} \quad E_r = \frac{3}{2} NkT, \text{ non-linear}$$

$$E_v = Nk \sum_{i=1}^{n'} \left(\frac{\Theta_{v,i}}{2} + \frac{\Theta_{v,i}}{(\exp(\Theta_{v,i} / T) - 1)} \right), \quad n'=3n-5 \text{ (linear or diatomic) or } 3n-6 \text{ (non-linear)}$$

$$\Theta_{v,i} = h \nu_i / k$$

Heat Capacity:

$$C_{V,t} = \frac{3}{2} Nk$$

$$C_{V,r} = Nk, \text{ linear or diatomic} \qquad C_{V,r} = \frac{3}{2} Nk, \text{ non-linear}$$

$$C_{V,v} = Nk \sum_{i=1}^{n'} \left[\left(\frac{\Theta_{v,i}}{T} \right)^2 \frac{\exp(\Theta_{v,i} / T)}{(\exp(\Theta_{v,i} / T) - 1)^2} \right], \quad n'=3n-5 \text{ (linear or diatomic) or } 3n-6 \text{ (non-linear)}$$

Entropy:

$$S_t = Nk \left[\frac{5}{2} + \ln \left[\left(\frac{2\pi k T \sum m_i}{h^2} \right)^{3/2} \frac{V}{N} \right] \right]$$

$$S_r = Nk \left(1 + \ln \frac{T}{\sigma \Theta_r} \right)$$

diatomic:

$$\Theta_r = h^2 / 8\pi^2 I k$$

$$S_r = Nk \left[\frac{3}{2} + \ln \frac{\pi^{1/2}}{\sigma} \left(\frac{T^3}{\Theta_A \Theta_B \Theta_C} \right)^{1/2} \right]$$

polyatomic:

$$\Theta_A \Theta_B \Theta_C = h^6 / [(8\pi^2 k)^3 I_x I_y I_z]$$

$$S_v = Nk \sum_{i=1}^{n'} \left[\frac{h \nu_i / kT}{e^{h \nu_i / kT} - 1} - \ln(1 - e^{-h \nu_i / kT}) \right], \quad n'=3n-5 \text{ (linear or diatomic) or } 3n-6 \text{ (non-linear)}$$