

I.

a) Find η_{TURBINE}

	<u>T</u>	<u>P</u>	<u>H</u>	<u>S</u>
Point 2	600°C	1000 kPa	3697.4	8.0292
Point 3	350°C	75 kPa	3176.1	8.5191
Isentropic Base Case 3	?	75 kPa	?	8.0292
Point 4	91.79	75 kPa	384.451	1.2131
Point 1	~91.79	1000 kPa	384.451 + $W_{s,\text{PUMP}}$	~1.2131

Interpolate for Isentropic base case at point 3

<u>T</u>	<u>S</u>	<u>H</u>	
200	7.9697	2876.6	$\frac{8.0292 - 7.9697}{8.0712 - 7.9697} = \frac{H_3' - 2876.6}{2925.8 - 2876.6}$
?	8.0292	?	
225	8.0712	2925.8	$H_3' = 2905.4$

$$W_{s,\text{TURBINE}} = 3176.1 - 3697.4 = -521.3$$

$$W_{s',\text{TURBINE}} = 2905.4 - 3697.4 = -792.0$$

$$\boxed{\eta_{\text{TURBINE}} = \frac{521.3}{792} = 0.658}$$

b) Find η_{cycle}

Need $W_{s,\text{NET}}$; $W_{s,\text{PUMP}}$ & Q_H

$$W_{s,\text{PUMP}} = V^L (1000 - 75) = 1.037 \frac{\text{cm}^3}{\text{gr}} (925 \text{ kPa}) = 959 \frac{\text{cm}^3 \cdot \text{kPa}}{\text{gr}}$$

$$959 \frac{\text{cm}^3 \cdot \text{kPa}}{\text{gr}} = \frac{959 \times 10^3 \text{ cm}^3 \cdot \text{kPa}}{\text{kg}} \times \frac{8.3145 \text{ J/mole} \cdot \text{K}}{8314 \text{ cm}^3 \cdot \text{kPa/mole} \cdot \text{K}} \times \frac{1000 \text{ J}}{\text{J}} = 0.959 \frac{\text{kJ}}{\text{kg}}$$

$$W_{s,\text{NET}} = -521.3 + 0.959 = \boxed{-520.34}$$

$$H_1 - H_4 = 0.959$$

$$H_1 = 0.959 + 384.451$$

$$H_1 = 385.41$$

$$Q_H = 3697.4 - 385.41$$

$$Q_H = 3312.0$$

$$\eta_{\text{cycle}} = \frac{520.34}{3312.0} = 0.157$$

II.

First guess, $P_2 = \sqrt{20 \times 500} = 100 \text{ psia}$

$$W_{s1}' = 132 - 113 = 19$$

$$W_{s1} = 19 / 0.75 = 25.33$$

$$H_2 = 113 + 25.3 = 138.3$$

$$W_{s2}' = 126 - 109 = 17$$

$$W_{s2} = 22.7$$

$$W_{s, \text{TOTAL}} = 48.0$$

Second guess

$$P_2 = 150 \text{ psia}$$

$$W_{s1}' = 138 - 113 = 25$$

$$W_{s2}' = 118 - 107 = 11$$

$$W_{s, \text{TOTAL}} = 36 / 0.75 = 48$$

Third guess $P_2 = 80 \text{ psia}$

$$W_{s1}' = 130 - 113 = 17$$

$$W_{s2}' = 130 - 110 = 20$$

$$W_{s, \text{TOTAL}} = 37 / 0.75 = 49.3$$

$$T_2 = 255^\circ \text{F}$$

$$P_2 = 100 \text{ psia}$$

$$T_3 = 110^\circ \text{F}$$

$$P_3 = 100 \text{ psia}$$

$$T_4 = 278^\circ \text{F}$$

$$P_4 = 500 \text{ psia}$$

Clearly, little difference. Use $P_2 = 100 \text{ psia}$

For $P_2 = 100 \text{ psia}$

$$W_{S1} + W_{S2} = 48.0 \text{ Btu/lbm}$$

b) heat load = $\Delta H_{2 \rightarrow 3}$

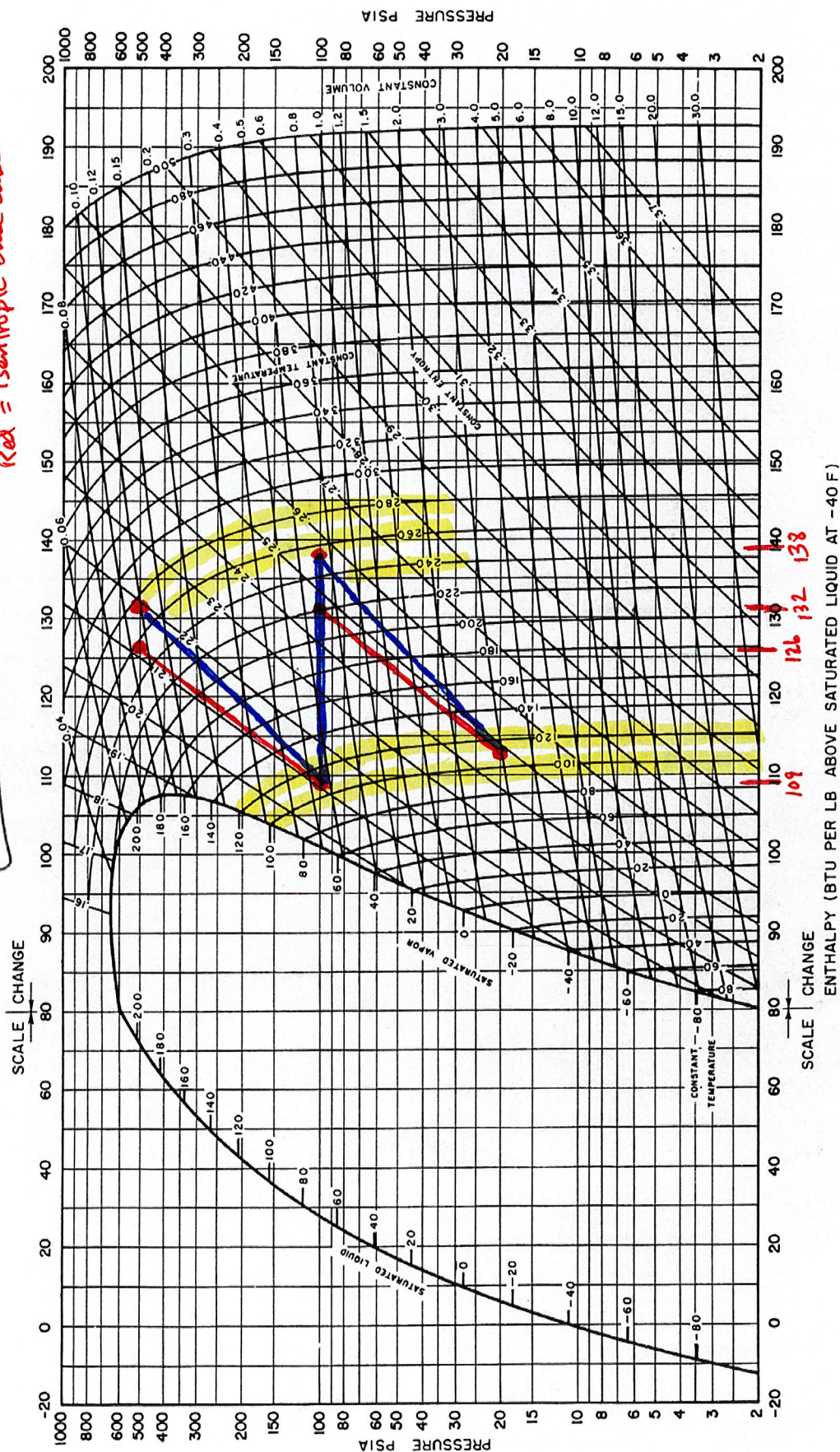
$$H_{2, \text{actual}} = 138.3$$

$$H_{2, \text{ideal}} = 109$$

$$\Delta H = Q = 29.3 \text{ Btu/lbm}$$

R-500

Blue = actual
Red = isentropic base case



TEMPERATURE IN °F VOLUME IN CU FT PER LB ENTROPY IN BTU PER LB PER °F

Fig. 9-11. Pressure-enthalpy diagram for refrigerant R-500. Note that 0 of enthalpy scale is taken at -40°F (-40°C). Also note that enthalpy scale changes at 80 Btu per lb. to help technician read superheat values.